

Subject	Year	Term	
Science	11	1	
Tonic			

11C1 Structure and Bonding

Content (Intent)

Prior Learning (Topic) 7C1 Physical changes & the particle model (lessons 1-8 and 10-12) 7C2 Atoms, Elements & Mixtures (lesson 1) 8C1 - properties of elements (lesson 1, 4, 5, 10 and 13), C1 Atomic structure and the periodic table (lesson 1, 5 and 13)

Ionic bonding - describe how ionic compounds are formed and draw dot and cross diagrams.

Ionic compounds – explain the properties of ionic compounds in terms of their bonding and structure **Covalent bonding** - describe covalent bonding and draw dot and cross diagrams.

Covalent compounds – explain the properties of in terms of their bonding and structure (for both simple molecular and giant covalent substances)

Polymers - explain the properties of polymers.

Metals and alloys – describe metallic bonding and explain the properties of metals and their alloys in terms of structure and bonding.

Graphene and fullerenes - explain the properties of graphene in terms of its structure and bonding.

Future Learning (Topic) A level chemistry OCR A 2.2 Electrons, bonding and structure and 3.1 The Periodic table.

How will knowledge and skills be taught?	How will your understanding be
(Implementation)	assessed & recorded (Impact)
Practical work	- 2 x standard homeworks (Level given.
Investigating the properties of ionic compounds	Written feedback. Response expected.)
Written Worksheets 11C3 lesson 1-7. Explanations of the properties of different substances in terms of their structure and bonding .	-Covalent dot and cross diagram: in exercise book. Marked and feedback givenDiagram of metallic bonding: in exercise book. Marked and feedback given.

How can parents help at home?

Look at the topic specific resources on the VLE

Use appropriate youtube channels: cognito, primrosekitten, khan academy, freesciencelessons.

Take an interest! Ask your children what they have learnt and be curious about their learning.

Helpful further reading/discussion

Reading	Vocabulary Lists	Careers Links
	Covalent	Material scientist.
	Ionic	
	Metallic	
	Giant structure	
	Lattice	
	Small molecules	
	Intermolecular forces	
	Delocalised electrons	
	Graphene	